Nickel Powder Type 123

Product Information

Nickel Powder Type 123

Nickel Powder Type 123 is used in alloying; and sintered powdered metallurgical operations.

Vale Canada Limited
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Suite 1600, South Tower, PO Box 70
Toronto, Ontario, Canada, M5J 2K2

Chemtrec 24 hr Emergency No. 1-800-424-9300 WHMIS Classification: D2B

Hazardous Ingredients

<table>
<thead>
<tr>
<th>Hazardous Ingredients</th>
<th>Calculated Composition</th>
<th>C.A.S. No</th>
<th>Oral LD&lt;sub&gt;50&lt;/sub&gt; -rat</th>
<th>Exposure limit (TLV)&lt;sup&gt;1,2&lt;/sup&gt; -mg/m³</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nickel (Ni)</td>
<td>&gt;99%</td>
<td>7440-02-0</td>
<td>&gt;9000 mg/kg</td>
<td>1.5*</td>
</tr>
</tbody>
</table>

*As inhalable fraction

Physical Data

Grey, odourless powder with an average particle size of 3 to 7 microns.

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>Mol. Wt.</th>
<th>Specific Gravity</th>
<th>Melting Point (°C)</th>
<th>Boiling Point (°C)</th>
<th>Solubility in H₂O (g/100ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nickel</td>
<td>58.71</td>
<td>8.9</td>
<td>1453</td>
<td>2732</td>
<td>0</td>
</tr>
</tbody>
</table>

Fire or Explosion Data

Metal powders heat treated in reducing atmospheres may become spontaneously combustible.

Reactivity Data

Like other metals, nickel can react with acids to liberate hydrogen gas which can form explosive mixtures in air.

Like other metal powders, nickel powder can react explosively or incandescently with substances such as ammonium nitrate, perchlorates, phosphorous, selenium, sulfur, etc.

Under special conditions nickel can react with carbon monoxide in reducing atmospheres to form nickel carbonyl, Ni(CO)<sub>4</sub>, a toxic gas.
Toxicological Properties

Nickel

Acute Toxicity:

a) Oral: Non toxic - LD$_{50}$ ORAL RAT >9000 mg/kg
b) Inhalation: No information available
c) Dermal: No information available

Corrosivity/Irritation:

a) Respiratory Tract: None
b) Skin: See sensitization section.
c) Eyes: Mechanical irritation may be expected.

Sensitization:

a) Respiratory tract: Nickel metal induced asthma is very rare. 3 case reports are available; the data is not sufficient to conclude that nickel metal is classified as a respiratory sensitizer.
b) Skin: Nickel metal is a well-known skin sensitizer. Direct and prolonged skin contact with metallic nickel may induce nickel allergy and elicit nickel allergic skin reactions in those people already sensitized to nickel, so called nickel allergic contact dermatitis.
c) Pre-existing conditions: Individuals known to be allergic to nickel should avoid contact with nickel whenever possible to reduce the likelihood of nickel allergic contact dermatitis reactions (skin rashes). Repeated contact may result in persistent chronic palmar/hand dermatitis in a smaller number of individuals, despite efforts to reduce or avoid nickel exposure.

Chronic toxicity:

a) Oral: No information available
b) Inhalation: Animal studies (rats) show that repeated dose inhalation of nickel damages the lung. Chronic inflammation, lung fibrosis and accumulation of nickel particles were observed.
c) Dermal: Direct and prolonged skin contact with nickel metal may cause nickel sensitization resulting in nickel allergic contact dermatitis /skin rash.

Mutagenicity / Reproductive toxicity: No data

Carcinogenicity:

a) Ingestion: The U.S. National Institute for Occupational Safety and Health (NIOSH) concluded that there is no evidence that nickel metal is carcinogenic when ingested.
b) Inhalation: There is limited information available from inhalation and intratracheal studies in animals. The U.S. National Toxicology Program has listed metallic nickel as reasonably anticipated to be a human carcinogen. To date, there is no evidence that nickel metal causes cancer in humans based on epidemiology data from workers in the nickel producing and nickel consuming industries.
The International Agency for Research on Cancer (IARC)(Vol 49) found there was inadequate evidence that metallic nickel is carcinogenic to humans but since there was sufficient evidence that it is carcinogenic to animals, IARC concluded that metallic nickel is possibly carcinogenic to humans (Group 2B). In 1997, the ACGIH categorized elemental nickel as: A5 "Not Suspected as a Human Carcinogen". Epidemiological studies of workers exposed to nickel powder and to dust and fume generated in the production of nickel alloys and of stainless steel have not indicated the presence of a significant respiratory cancer hazard.

**Preventative Measures**

Do not inhale powder. Keep container closed when not in use. Ventilation is normally required when handling or using this product to keep exposure to airborne nickel below the exposure limit. If ventilation alone cannot so control exposure, use NIOSH-approved respirators selected according to the current edition of the Selection, Care and Use of Respirators CSA Z94.4. Maintain airborne nickel levels as low as possible.

Avoid repeated skin contact. Wear suitable gloves. Wash skin thoroughly after handling. Launder clothing and gloves as needed. Do not store near acids or reactive substances.

If spilled, collect spills by wet sweeping or by vacuuming with the vacuum exhaust passing through a high efficiency particulate arresting (HEPA) filter if the exhaust is discharged into the workplace. Wear appropriate NIOSH-approved respirators if collection and disposal of spills is likely to cause the concentration of airborne contaminants to exceed the exposure limits.

**First Aid Measures**

If exposure to nickel carbonyl is suspected, seek medical attention immediately. For skin rashes, seek medical attention. Cleanse wounds thoroughly to remove any particles.

**Personal Protection**

Maintain airborne levels as low as possible.

**Disposal Considerations**

Nickel-containing waste is normally collected to recover nickel values. Should waste disposal be deemed necessary, follow local relevant governmental regulations.
Preparation Information

Prepared by:

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Note:
Vale Canada believes that the information in this Material Safety Data Sheet is accurate. However, Vale Canada makes no express or implied warranty as to the accuracy of such information and expressly disclaims any liability resulting from reliance on such information.

Footnotes:
1. Threshold Limit Value of the American Conference of Governmental Industrial Hygienists.
2. Exposure Limits for user operations will depend on the relevant governmental regulations.
3. Describes possible health hazards of the product supplied. If user operations change it to other chemical forms, whether as end products, intermediates or fugitive emissions, the possible health hazards of such forms must be determined by the user.