PTI Nickel Matte

**Product Information**

PTI Nickel Matte

PTI Matte is used for production of nickel containing materials.

*Manufactured by:*  
PT International Nickel Indonesia Tbk  
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Emergency Tel: +62 21 524 9999 (Soroako Site)

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*Imported by:*  
Vale Inco Japan Limited  
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MORI Tower 25F  
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Minatoku,  
Tokyo 105-6225, Japan

**Hazards Identification**

**GHS:**

<table>
<thead>
<tr>
<th>Health</th>
<th>Environmental</th>
<th>Physical</th>
</tr>
</thead>
<tbody>
<tr>
<td>Skin Sensitization – Category 1</td>
<td>Aquatic Toxicity – Acute 1</td>
<td>------</td>
</tr>
<tr>
<td>Carcinogenicity – Category 1</td>
<td>Aquatic Toxicity – Chronic 1</td>
<td>------</td>
</tr>
<tr>
<td>STOT * Repeated Exposure – Category 1</td>
<td>---</td>
<td>------</td>
</tr>
</tbody>
</table>

* - Single Target Organ Toxicity

Symbols: Exclamation mark, Health Hazard, Environmental

Signal Word: Danger

Hazard Statements:

- May cause an allergic skin reaction.
- Causes damage to lungs through prolonged or repeated inhalation exposure
- May cause cancer by inhalation
- Very toxic to aquatic life with long lasting effects
Precautionary Statements:

Prevention:
Obtain special instruction before use
Do not handle until all safety precautions have been read and understood
Use personal protective equipment as required
Avoid breathing dust or fume.
Contaminated work clothing should not be allowed out of the workplace.
Wear protective gloves and protective clothing
Wash hands, and face thoroughly after handling.
Avoid release to the environment

Response:
If exposed or concerned: get medical advice/attention
If on skin: Wash with plenty of soap and water.
If skin irritation or rash occurs: Get medical advice/attention.
Wash contaminated clothing before reuse
See First Aid section for specific treatment information
Get medical advice/attention if you feel unwell.
Collect spillage

Storage:
Store locked up

Disposal:
Dispose of contents/container in accordance to local/regional/national/international regulations

Composition

<table>
<thead>
<tr>
<th>Substance</th>
<th>Mixture</th>
</tr>
</thead>
</table>

<table>
<thead>
<tr>
<th>Hazardous Ingredients</th>
<th>Typical Composition (%)</th>
<th>C.A.S. Number</th>
</tr>
</thead>
<tbody>
<tr>
<td>Nickel subsulfide</td>
<td>60-80</td>
<td>12035-72-7</td>
</tr>
<tr>
<td>Nickel cobalt iron alloy</td>
<td>balance</td>
<td>N/A</td>
</tr>
<tr>
<td>Cobalt (II) sulfide</td>
<td>&lt;1</td>
<td>1317-42-6</td>
</tr>
</tbody>
</table>

First Aid Measures

Ingestion:
If swallowed do not induce vomiting. Seek medical attention.

Inhalation:
If inhaled, remove from contaminated area. Seek medical attention.

Skin:
Wash thoroughly with water. For rashes seek medical advice. Show label if possible.

Eyes:
Irrigate eyeball thoroughly with water for at least 10 minutes. If discomfort persists seek medical attention.
Wounds: Cleanse thoroughly to remove any particles.

Fire Fighting Measures

Suitable extinguishing media: Any, type to be selected according to materials stored in the immediate area.

Special risks: Non-flammable. Keep containers cool with water spray.

Special protective equipment for fire fighting: None needed. Wear protective equipment if required for other materials within the immediate vicinity.

Accidental Release Measures

Person related precautionary measures: Avoid generation of dusty atmospheres. Do not inhale dusts. Wear protective clothes, gloves, safety glasses, and respirator.

Environmental Protection measures: Contain spillage. Avoid entry in sewers, and water streams.

Procedures for cleaning/absorption: Pick up and replace in original container. Nickel-containing material is normally collected to recover nickel values.

Handling and Storage

Handling: Prevent the generation of inhalable dusts e.g. by the use of suitable ventilation. Do not inhale dust. Wear appropriate nationally approved respirators if handling is likely to cause the concentration limits of airborne nickel to exceed the locally prescribed exposure limits. Wear suitable protective clothing and gloves.

Storage: Keep in the container supplied, and keep container closed when not in use. Store locked up. Local regulations should be followed regarding the storage of this product.
Exposure Controls / Personal Protection

Nickel Subsulfide – CAS 12035-72-7

<table>
<thead>
<tr>
<th></th>
<th>Exposure Limit</th>
</tr>
</thead>
<tbody>
<tr>
<td>ACGIH TLV-TWA</td>
<td>0.1 mg/m$^3$ * (as Ni)</td>
</tr>
<tr>
<td>UK WEL</td>
<td>0.5 mg/m$^3$ (as Ni)</td>
</tr>
<tr>
<td>Japan</td>
<td>1 mg/m$^3$ (as Ni)</td>
</tr>
<tr>
<td>Korea</td>
<td>2 mg/m$^3$ (as Ni)</td>
</tr>
<tr>
<td>Indonesia</td>
<td>1 ppm</td>
</tr>
</tbody>
</table>

* - in inhalable fraction

Cobalt (II) sulfide – CAS 1317-42-6

<table>
<thead>
<tr>
<th></th>
<th>Exposure Limit</th>
</tr>
</thead>
<tbody>
<tr>
<td>ACGIH TLV-TWA</td>
<td>0.02 mg/m$^3$ (as Co)</td>
</tr>
<tr>
<td>UK WEL</td>
<td>0.1 mg/m$^3$ (as Co)</td>
</tr>
<tr>
<td>Japan</td>
<td>0.05 mg/m$^3$ (as Co)</td>
</tr>
<tr>
<td>Korea</td>
<td>0.05 mg/m$^3$ (as Co)</td>
</tr>
</tbody>
</table>

Maintain airborne levels as low as possible.

**Occupational exposure controls:**

a. **Respiratory protection:** Ventilation may be required if user operations change it to other physical or chemical forms, whether as end products, intermediates or fugitive emissions, which are inhalable.

b. **Eye protection:** None

c. **Hand & Skin Protection:** Avoid repeated skin contact. Wear suitable protective clothing and gloves, which should be selected specifically for the working place, depending on concentration and quantity of the hazardous material (overalls and leather/rubber gloves). Wash skin thoroughly after handling and before eating, drinking or smoking. Change contaminated clothing frequently. Launder clothing and gloves as needed. Use of skin-protective barrier cream advised.

**Physical and Chemical Properties**

Black-grey colored granules; insoluble in water
<table>
<thead>
<tr>
<th>Property</th>
<th>Value</th>
</tr>
</thead>
<tbody>
<tr>
<td>pH</td>
<td>Not Applicable (N/A)</td>
</tr>
<tr>
<td>Boiling point/boiling range</td>
<td>2582°C</td>
</tr>
<tr>
<td>Melting point</td>
<td>540°C</td>
</tr>
<tr>
<td>Flash Point</td>
<td>N/A</td>
</tr>
<tr>
<td>Evaporation rate</td>
<td>N/A</td>
</tr>
<tr>
<td>Vapor Temperature</td>
<td>2857°C</td>
</tr>
<tr>
<td>Flammability</td>
<td>N/A</td>
</tr>
<tr>
<td>Explosive properties</td>
<td>Not explosive</td>
</tr>
<tr>
<td>Vapor pressure</td>
<td>N/A</td>
</tr>
<tr>
<td>Vapor density</td>
<td>N/A</td>
</tr>
<tr>
<td>Relative density</td>
<td>No data</td>
</tr>
<tr>
<td>Solubility cold water</td>
<td>Insoluble</td>
</tr>
<tr>
<td>Solubility hot water</td>
<td>Insoluble</td>
</tr>
<tr>
<td>Partition coefficient</td>
<td>N/A</td>
</tr>
<tr>
<td>Auto-ignition temperature</td>
<td>N/A</td>
</tr>
<tr>
<td>Decomposition temperature</td>
<td>N/A</td>
</tr>
<tr>
<td>Oxidizing properties</td>
<td>Not oxidizing</td>
</tr>
<tr>
<td>Viscosity</td>
<td>N/A</td>
</tr>
<tr>
<td>Packaged Density</td>
<td>3.8 g/cm³</td>
</tr>
<tr>
<td>Loose Density</td>
<td>3.3 g/cm³</td>
</tr>
<tr>
<td>Particle size</td>
<td>70% 0.15 mm – 0.8 mm</td>
</tr>
<tr>
<td></td>
<td>Max. 10% &gt;0.8 mm</td>
</tr>
<tr>
<td></td>
<td>Max. 20% &lt;0.15mm</td>
</tr>
</tbody>
</table>

**Stability and Reactivity**

*Conditions to be avoided:* No hazardous exothermic reaction.

*Substances to be avoided:* Like other metals, Nickel may react with acids to liberate hydrogen gas, which may form explosive mixtures in air.

*Hazardous decomposition products:* No information available.

**Toxicological Information**

As a mixture the toxicological properties of this product are unknown. The toxicology of the reported ingredients is summarized below.

**Nickel Subsulfide**

LD50 ORAL RAT > 5000 mg/kg
Inhalation: The National Toxicology Program has listed nickel subsulfide as reasonably anticipated to be a carcinogen based on the production of injection-site tumors. The International Agency for Research on Cancer (IARC) concluded there was sufficient evidence that nickel compounds are carcinogenic to humans and that crystalline nickel sulfides are carcinogenic to animals. Epidemiological studies of workers engaged in the oxidation of nickel subsulfide (Ni₃S₂) by dusty processes indicated the presence of a significant respiratory cancer hazard.

The American Conference of Governmental Industrial Hygienists (ACGIH) have classified nickel subsulfide as a Confirmed Human Carcinogen; category A1.

Rats exposed by inhalation to ~1 mg Ni₃S₂/m³ experienced an increased incidence of malignant lung tumors. Repeated intratracheal instillation of nickel subsulfide produced an increased incidence of malignant lung tumors in rats. Repeated intratracheal instillation of nickel subsulfide did not produce an increased incidence of malignant lung tumors in hamsters when administered at the maximum tolerated dose.

Wounds: Nickel subsulfide is a potent experimental carcinogen in rodents by parenteral routes of administration.

Ingestion: The U.S. National Institute for Occupational Safety and Health (NIOSH) concluded there is no evidence that nickel and its inorganic compounds are carcinogenic when ingested.

Nickel cobalt iron alloy
A literature search found no toxicological information for Nickel cobalt iron alloy. Toxicology information is provided for its main constituents; nickel, cobalt, and iron.

Nickel
Acute Toxicity:
 a) Oral: Non toxic - LD50 ORAL RAT >9000 mg/kg
 b) Inhalation: No information available
 c) Dermal: No information available.

Corrosivity/Irritation:
 a) Respiratory Tract: None
 b) Skin: See sensitization section.
 c) Eyes: Mechanical irritation may be expected.

Sensitization:
 a) Respiratory tract: Nickel metal induced asthma is very rare. 3 case reports are available; the data is not sufficient to conclude that nickel metal is classified as a respiratory sensitizer.
 b) Skin: Nickel metal is a well-known skin sensitizer. Direct and prolonged skin contact with metallic nickel may induce nickel allergy and elicit nickel allergic skin reactions in those people already sensitized to nickel, so called nickel allergic contact dermatitis.
 c) Pre-existing conditions: Individuals known to be allergic to nickel should avoid contact with nickel whenever possible to reduce the likelihood of nickel allergic contact dermatitis reactions (skin
rashes). Repeated contact may result in persistent chronic palmar/hand dermatitis in a smaller number of individuals, despite efforts to reduce or avoid nickel exposure.

**Chronic toxicity:**

a) **Oral:**
No information available

b) **Inhalation:**
Animal studies (rats) show that repeated dose inhalation of nickel damages the lung. Chronic inflammation, lung fibrosis and accumulation of nickel particles were observed.

c) **Dermal:**
Direct and prolonged skin contact with nickel metal may cause nickel sensitization resulting in nickel allergic contact dermatitis /skin rash.

**Mutagenicity / Reproductive toxicity:**
No data.

**Carcinogenicity:**

a) **Ingestion:**
The U.S. National Institute for Occupational Safety and Health (NIOSH) concluded that there is no evidence that nickel metal is carcinogenic when ingested.

b) **Inhalation:**
There is limited information available from inhalation and intratracheal studies in animals. The U.S. National Toxicology Program has listed metallic nickel as reasonably anticipated to be a human carcinogen. To date, there is no evidence that nickel metal causes cancer in humans based on epidemiology data from workers in the nickel producing and nickel consuming industries.

The International Agency for Research on Cancer (IARC)(Vol 49) found there was inadequate evidence that metallic nickel is carcinogenic to humans but since there was sufficient evidence that it is carcinogenic to animals, IARC concluded that metallic nickel is possibly carcinogenic to humans (Group 2B). In 1997, the ACGIH categorized elemental nickel as: A5 "Not Suspected as a Human Carcinogen". Epidemiological studies of workers exposed to nickel powder and to dust and fume generated in the production of nickel alloys and of stainless steel have not indicated the presence of a significant respiratory cancer hazard.

**Cobalt**

LD50 oral rat: 6171 mg/kg

**Inhalation:**
Asthmatic symptoms and pulmonary fibrosis occurring in the tungsten carbide industry may be related to the inhalation of metallic cobalt dust. Evidence of polycythemia (an increase in the total red cell mass of the blood in the body) and altered thyroid, kidney and liver function have also been found. Excessive doses of metallic cobalt have produced cardiac changes in miniature swine.

**Skin Contact:**
Repeated contact with metallic cobalt can cause cobalt sensitivity and allergic skin rashes.

**Wounds:**
Cobalt powders have caused tumors at the site of injection in rodents. However, studies of cobalt containing prostheses do not suggest a significant risk for humans.

**Pre-existing Conditions:**
Sensitized individuals may experience an allergic skin rash or asthma.

**Iron**
LD50 oral rat: 30,000 mg/kg

Prolonged eye contact with the metal dust could cause rust-brown coloured spots forming around the particles and if left for several years, permanent damage could result.

**Cobalt (II) Sulfide**
Oral LD50 >5000 mg/kg

A literature search found no toxicological information for cobalt sulphide. Toxicology is expected to be similar to cobalt oxide.

**Cobalt Oxide**
Inhalation: Some workers engaged long-term in the production of cobalt oxides showed symptoms of chronic bronchitis. Inhalation experiments show that cobalt oxide accumulated in the lymph nodes of dogs suggesting it is tissue insoluble.

**Ecological Information**

Very toxic to aquatic organisms may cause long-term adverse effects in the aquatic environment. Do not discharge into sewer or waterways.

**Disposal Considerations**

Nickel-containing material is normally collected to recover nickel values. Should disposal be deemed necessary, follow local regulations.
Transport Information

| International Maritime Dangerous Goods Code | Not regulated. After January 1, 2010 this material will be considered a Dangerous Good for Transport. The proper shipping name will be: UN3077, Environmentally Hazardous Substance, Solid, NOS (Nickel powder), class 9, PG III |
| U.S. Dept. of Transportation Regulations | Not regulated. |
| Canadian Transportation of Dangerous Goods Act | Not regulated. |
| European Agreement Concerning the International Carriage of Dangerous Goods by Road | Not regulated. After July 1, 2009 this material will be considered a Dangerous Good for Transport. The proper shipping name will be: UN3077, Environmentally Hazardous Substance, Solid, NOS (Nickel powder), class 9, PG III |

Regulatory Information

Other Information

Safety Data Sheet prepared by:

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Note:
**Vale Inco believes that the information in this Material Safety Data Sheet is accurate. However, Vale Inco makes no express or implied warranty as to the accuracy of such information and expressly disclaims any liability resulting from reliance on such information.**

1. Threshold Limit Values of the American Conference of Governmental Industrial Hygienists. 2008.